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2 **Diurnal fluctuation of root oxygen release rate and dissolved**
3 **oxygen budget in wetland mesocosm**

4

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17

18 **Abstract**

19 To assess the role of plants for oxygen offering in constructed wetlands, this study

20 _____

21 DO is dissolved oxygen, CW is constructed wetlands, HLR is hydraulic load rate, PAR

22 is photosynthetically active radiation.

24 experimentally evaluated the amount of root oxygen release rate using mass balance
25 method. The mass balance calculation is based on the following components:
26 respiratory oxygen consumption of the roots; oxygen required for degradation of the
27 organic matters; oxygen required for nitrification; and the amount of dissolved oxygen
28 in the influent, effluent and substrate water. Experimental results have demonstrated
29 that the root oxygen release rate was ranged from 20.3 to 58.3gO₂/m².d with average
30 value of 38.4gO₂/m².d, which was affected dramatically by light intensity. Only 35%
31 and 9% of the oxygen released by roots were used in the degradation of organic
32 matters and nitrogen-nitrification, respectively, while 56% was used for roots
33 respiration with little to be released to the surrounding medium. The results also
34 showed that diurnal fluctuation of oxygen release and light intensity followed
35 unimodal distribution. Meanwhile, a better understanding of the DO (dissolved oxygen)
36 budget was proposed. Root oxygen release rate could be described by two fractions.
37 One is “net specific oxygen release rate” and the other is “dischargeable oxygen
38 release rate”.

39 *Keywords:* Constructed wetlands; Diurnal variation; DO budget; Mass balance method;
40 Oxygen release rate

41

42 **1. Introduction**

43 The global use of constructed wetlands (CW) as an eco-friendly, low maintenance,
44 low-cost, and efficient alternative to conventional wastewater treatment has been
45 growing over the last decades [1,2]. Pollutants are reduced from wastewater in the CW

46 by a variety of physiochemical and biological processes. Available oxygen in CW is an
47 important factor in degradation of organic matter and transformation of
48 ammonium-nitrogen, both of which are oxygen limiting processes. When hypoxia
49 (dissolved oxygen <2mg/L) occurs, a wide range of negative effects on aquatic
50 ecosystems and organisms could be caused [3].

51 Coupled nitrification-denitrification is considered as the main N sink in CW [4].
52 Nitrification is an aerobic chemoautotrophic process while denitrification is an
53 anaerobic heterotrophic microbial process. The nitrification step represents the main
54 limiting factor for N removal in CW because of low oxygen availability in most cases
55 [5,6]. Low oxygen content results in low aerobic organic matter decomposition [6,7].
56 In order to enhance N removal efficiency, oxygen must be provided to the nitrifying
57 microbes through oxygenation of the wetland matrix via several means. Plants provide
58 oxygen to the rhizosphere via passive or active oxygen transport through their stems
59 from the atmosphere to the roots resulting in higher N removal rates in planted CW [8].

60 Recently, the interest in the main term of oxygen is increasing. Many studies have
61 been conducted to illustrate the mechanisms responsible for mass balance of DO
62 (dissolved oxygen) in estuarine and coastal areas worldwide [9]. However, there is
63 little detailed information about the oxygen mass balance in CW. Furthermore, only
64 few attempts have yet been made to evaluate the diurnal changes of oxygen
65 consumption during degradation of organic matter or nitrogen oxidation. Accordingly,
66 little was known about the diurnal variation of oxygen release by the roots of wetland
67 plant. The role of plants for oxygen offering remains unclear. Therefore, the objective

68 of the present study was to provide a better understanding of the DO budget and the
69 daily variation of the root oxygen release rates, which were explored based on mass
70 balance method.

71

72 **2. Materials and methods**

73 *2.1. Experimental mesocosm*

74 The model wetland used in this study was a sub-surface vertical flow CW.
75 Mesocosm was prepared from perspex column of 150mm in diameter and 1000mm in
76 height. The mesocosm was filled with 2-5mm sized scoria as the substrate in a depth of
77 900 mm, giving an average porosity value of 0.46. Four sampling ports (S1, S2, S3,
78 S4), inlets and outlets at different heights of the mesocosm (model CW) were setup
79 (Fig. 1). Young *Acorus calamus* Linns collected from a natural wetland located in
80 Xuanwu Lake, Nanjing, China, were planted in the wetland mesocosm. It was exposed
81 to the open air with natural light outside the laboratory building.

82 Ammonium chloride, potassium dihydrogen phosphate and glucose were used as
83 the sources of ammonium and organic matters. They were mixed with tap water to
84 prepare the artificial wastewater [10]. The characteristics of the wastewater were as
85 follows: TN 15.1mg/L (ranged between 12.0-18.7mg/L), $\text{NH}_4^+\text{-N}$ 13.6 (ranged
86 between 12.1-15.2 mg/L), $\text{NO}_3^-\text{-N}$ 1.7 mg/L (ranged between 1.5-1.8 mg/L) and
87 $\text{NO}_2^-\text{-N}$ 0.006 mg/L (ranged between 0-0.011 mg/L). The wastewater was pumped into
88 a holding tank, from where the mesocosm was supplied. Hydraulic load rate (HLR)
89 was $0.16\text{m}^3/\text{m}^2$ d, and the hydraulic retention time was about 2.43d. The model CW

90 was operated for one year. When the mesocosm reached its steady state after the
91 start-up period, Samples of influent and effluent were collected from the six sampling
92 points along the mesocosm every two hours. The samples were then analysed for DO,
93 BOD₅, total nitrogen, ammoniacal-nitrogen, nitrite and nitrate according to the
94 Standard Methods for Examination of Water and Wastewater [11]. Light intensity was
95 measured every two hours using a luminometer (MODEL ZDS-10F-2D). The unit of
96 light intensity is lux. In this paper, light intensity was described as photosynthetically
97 active radiation (PAR). One lux is $0.019 \mu\text{mol}\cdot\text{m}^{-2} \text{ s}^{-1}$ [12].

98

99 *2.2. Respiration of wetland plants*

100 To prepare the examination, the surface of substrate (scoria) was covered with
101 vaseline to prevent re-aeration from the atmosphere. The plants were covered with a
102 black plastic bag to prevent photosynthesis. These were to make sure that the
103 respiration was mainly taking place in the substrate. For the purpose of examining the
104 respiration of wetland plants, the oxygen in the blank influent (without pollutants) of
105 the wetland mesocosm was carefully pre-monitored. Such the oxygen was then
106 absorbed by the root systems of plant in the process of respiration. Two hours later, the
107 oxygen in effluent was examined. The difference of oxygen (in mass) can be used to
108 calculate respiration rate.

109 The difficulty of the examination is the disturbance of oxygen from the air. The
110 details of the procedure were as follow: Firstly, the volume of the blank influent was
111 carefully controlled and the DO concentration was pre-monitored. Secondly, before the

112 blank influent was introduced to the mesocosm, the O₂ in the pore space of the
113 substrate was replaced by N₂. The influent was then put into the mesocosm suddenly.
114 Two hours later, the mesocosm was carefully drained using a vacuum-pump. At the
115 same time, N₂ was injected into the mesocosm to prevent re-aeration. Finally, the
116 volume and DO concentration of the effluent of the wetland mesocosm was examined,
117 this allows to calculating the O₂ mass.

118

119 2.3. Statistical analyses

120

121 2.3.1. Oxygen consumption rate calculations.

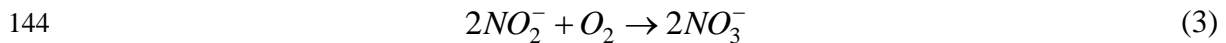
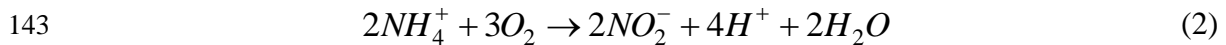
122 The reasonable way to calculate the DO consumed by bacteria is via the
123 decomposition of the organic matters [13]. Organic pollutants in wastewater are
124 adsorbed on microbial cell surface which inhabits on the substrate surface. Pollutants
125 are metabolized by microbes under catalysis of various enzymes. Oxygenolysis of
126 partial organic matter by microbes is called catabolic metabolism. The other organic
127 pollutants are taken in by microbes, and converted to new cells, this process is called
128 anabolism. Based on these two metabolic processes, the oxygen demand of reducing
129 organic matter ($O_{organic}$, mg/L) was analyzed theoretically as follows [13, 14]:

$$130 \quad O_{organic} = a' BOD_r + b' P \quad (1)$$

131 Where the first part is the oxygen consumed in the process of catabolic
132 metabolism. The second part is the oxygen consumed in the process of anabolism. a' is
133 oxygen demand for the complete degradation of 1kg BOD₅; BOD_r is the removal of

134 BOD₅ in substrate, mg/L; b' is the oxygen demand of active biofilm per unit weight; P
 135 is the quantity of active biofilm which adsorbs to every 1 m³ substrate (kg/m³). Oxygen
 136 consumption rate (gO₂/m²d) could be calculated as multiplying the oxygen demand by
 137 hydraulic load rate (0.16m³/m² d). It is noted that Ye [13] reported that $b'P$ value was
 138 to be 0-0.00042kg/m³ at different depths. Therefore, $b'P$ could be ignored. Here, the
 139 value of a' is 3.76 (kgO₂/kgBOD₅) since 3.53kg glucose is corresponding to 1kgBOD₅
 140 while 1kg glucose is corresponding to 1.07kg oxygen.

141 Oxygen consumed by nitrification was calculated based on Eq. (2) and Eq. (3).
 142 Oxidizing 1g of NH₄⁺-N and NO₂⁻-N requires 3.43g and 1.14g of oxygen, respectively.



146 2.3.2. Oxygen mass balance calculations.

147 In CW mesocosm, the root oxygen release could be assessed with mass balance
 148 method [15]. The main sources of oxygen in the substrate water include the inflow
 149 carrying, aeration and oxygen releasing from roots of plant. Output of oxygen includes
 150 outflow carrying, nitrification consumption and organic matter consumption. Mass
 151 balance equation of oxygen is given below:

$$152 \quad O_{plant} = (O_{out} + O_{organic} + O_{nitrification} + O_{substract} + O_{res}) - O_{in} \quad (4)$$

153 Where O_{plant} is root oxygen release, gO₂m⁻²d⁻¹; O_{out} is outflow carrying, gO₂m⁻²d⁻¹;
 154 $O_{organic}$ is organic matter consumption, gO₂m⁻²d⁻¹; $O_{nitrification}$ is nitrification
 155 consumption, gO₂m⁻²d⁻¹; $O_{substract}$ is DO in substrate water, gO₂m⁻²d⁻¹; O_{res} is

156 respiration consumption, $\text{gO}_2\text{m}^{-2}\text{d}^{-1}$; O_{in} is inflow carrying, $\text{gO}_2\text{m}^{-2}\text{d}^{-1}$.

157

158 *2.3.3. Prediction of root oxygen release.*

159 The root oxygen release rate in CW can be predicted using Gaussian function [16].

160 The light intensity, which was examined simultaneous with the water sample, can be

161 used for prediction purpose.

162

163 **3. Results**

164 Variations of oxygen consumption rate are illustrated in Fig. 2. It shows that the

165 oxygen consumption fluctuations for organic matter degradation and nitrification are in

166 the range of $3.1\text{-}26.1 \text{ gO}_2/\text{m}^2.\text{d}$ and $0.8\text{-}7.6 \text{ gO}_2/\text{m}^2.\text{d}$, respectively. More importantly,

167 the results have revealed a significant difference in the oxygen consumption rate

168 during day and night. Oxygen consumption rate was higher during daytime, following

169 an unimodal distribution pattern. The removal of organic matter and nitrogen is

170 accordingly followed the same pattern as oxygen consumption.

171 Fig. 3 illustrates the DO concentration of the water samples in the CW mesocosm,

172 which was calculated by combining several DO concentrations at different depths. The

173 CW mesocosm was divided into 5 units from the bottom at 0cm, 20cm, 45cm, 60cm,

174 and 75cm, respectively. The DO concentration and the volume of every unit were used

175 to calculate the total mass. The figure reveals that DO in the water changed in two

176 “day-night” cycle. In substrate water DO levels fluctuated, ranging from 0.03-0.39, due

177 to the oxygen released by root systems of wetland plant. Relatively, the inflow DO

178 concentration was stable. DO levels were dropped significantly in CW mesocosm,
179 from an average of 0.9mg/L in influent to 0mg/L in effluent.

180 Fig. 4 illustrates the daily changes of respiration rate by the roots of plant. Root
181 respiration measured as O₂ consumption in CW system is generally followed a daily
182 cycle. Respiration rate is ranged from 14.4 gO₂/m².d to 30.3 gO₂/m².d. Temperature is
183 ranged from 15°C to 35°C. The respiration rate is increased with increasing
184 temperature in the morning. However, the decreasing temperature made it decrease in
185 the afternoon. The results revealed that respiration of wetland plants was influenced by
186 temperature.

187 Root oxygen release rates were calculated by mass balance method and plotted in
188 Fig. 5. The mass balance has considered the following components: respiratory oxygen
189 consumption of the roots; oxygen required for degradation of the organic matters;
190 oxygen required for nitrification; and the amount of dissolved oxygen in the influent,
191 effluent and substrate water. The daily changes of PAR were also examined. Root
192 oxygen release rate is ranged from 20.3 to 58.3gO₂/m².d. The peak value of oxygen
193 release rate (54.0-58.3gO₂/m².d) was observed during the daytime at 15:00 hrs while
194 the maximum light intensity was observed at 13:00 hrs. Obviously, the peak of root
195 oxygen release appeared after the peak of light intensity. Light intensity data have been
196 demonstrated to predict the root oxygen release rate via the Gaussian function [16],
197 which is in the form of follows:

$$198 \quad V_O = ae^{-\frac{(t-t_{Omax})^2}{c^2}} \quad (5)$$

199 Where t is time (4:00am~20:00pm); a (in Gaussian function) is the maximum

200 value of oxygen release rate in a whole day; t_{Omax} is the location of the symmetry axis
201 in Gaussian function; c expresses the gradient of Gaussian function. Decrease in c is
202 relative to steep Gaussian function while increase in c is relative to gentle Gaussian
203 function.

204 Light intensity data during daytime (4:00am-20:00pm) also follow Gaussian
205 function (Fig. 5(b)), it can be described as:

$$206 \quad PAR = be^{-\frac{(t-t_{Lmax})^2}{d^2}} \quad (6)$$

207 Where PAR is the photosynthetically active radiation, $\mu\text{mol}\cdot\text{m}^{-2}\text{ s}^{-1}$; t is time; b is
208 the peak value of PAR in a whole day; and d is the gradient of unimodal.

209 The Gaussian parameters calculated by the light intensity data of the current study
210 are: $b_1=1094$, $d_1=3.477$; $b_2=1465$, $d_2=3.47$; $a_1=249.00$, $c_1=3.59$; $a_2=467.85$, $c_2=3.58$.
211 Then, the root oxygen release rate was predicted and jointly illustrated in Fig. 5. The
212 predicted values were slightly lower than the mass balance results. Oxygen release is
213 increased gradually with increasing light intensity in the morning. However, a decrease
214 in the oxygen release rates occurred following the decreased light intensity in the
215 afternoon. It appeared a large difference of root oxygen release for mass balance results
216 and predicted values at night. These variations indicate a significant time dependent
217 interval for oxygen release by plants during day and night.

218

219 **4. Discussion**

220 Based on the oxygen source and sink in the CW mesocosm, a mass balance for
221 oxygen was calculated. Regarding the consideration of the oxygen sources, it should be

222 pointed out that the amount of reaeration in subsurface CW was ignored in the mass
223 balance calculations because the levels were generally low. As described by Horn and
224 Smucker [17], oxygen diffusion is limited by partial or complete water saturation.
225 Accordingly, the anoxic sites may be developed. The influent DO was $0.9\text{gO}_2/\text{m}^2\cdot\text{d}$ in
226 average (Fig. 3). With regard to the oxygen sink, the average amount of oxygen
227 required for degradation of the organic and nitrogen-nitrification was $13.8\text{gO}_2/\text{m}^2\cdot\text{d}$
228 and $3.4\text{gO}_2/\text{m}^2\cdot\text{d}$, respectively (Fig. 2). The dissolved oxygen in the effluent and
229 substrate water was respectively $0\text{gO}_2/\text{m}^2\cdot\text{d}$ and $0.2\text{gO}_2/\text{m}^2\cdot\text{d}$ (Fig. 3). The respiratory
230 oxygen consumption of the roots and rhizomes was almost $21.9\text{gO}_2/\text{m}^2\cdot\text{d}$ (Fig. 4).
231 Finally, the oxygen released by root systems of wetland plant could then be obtained,
232 which was $38.4\text{gO}_2/\text{m}^2\cdot\text{d}$. Such the amount of oxygen release rate could be termed as
233 “net specific oxygen release rate”.

234 The DO budget showed that only 35.0% and 8.7% of the oxygen released by roots
235 were used in the degradation of organic matters and nitrogen-nitrification through
236 heterotrophy bacterial. 55.8% was used for roots respiration, this is because the oxygen
237 released from one part of the root system was reabsorbed by another. The respiratory
238 oxygen consumption of the roots could almost balance the oxygen release, leaving
239 little to be released to the surrounding medium. Such the amount of oxygen release rate
240 could be termed as “dischargeable oxygen release rate”. Similar responses/patterns
241 have been reported in Eldridge and Morse [18], who showed that the rate of oxygen
242 consumption exceeds its rate of input from photosynthetic generation.

243 It is interesting that the results of DO budget could be used to explain the debate

244 about the quantity of oxygen released by root systems of wetland plant. It is noted from
245 Bezbaruah and Zhang [19] that plants did not release enough oxygen to meet the total
246 oxygen demand of bulk wastewater. The maximum oxygen release rate of only
247 11.0-12.5mgO₂/kgwet.h (0.34-0.39 μmol O₂/gDW_{root}.h) was reported by Soda et al.
248 [20]. These results can be treated as the “dischargeable oxygen release rate”. Since the
249 root oxygen release rate was monitored in oxygen-scavenging solution (or wastewater)
250 using DO microelectrode. The oxygen released from roots was reabsorbed due to the
251 respiration. Contrarily, the obviously higher oxygen release rate of
252 148-798μmolO₂/gDW_{root}.h for *M. spicatum* and *P. crispus* was reported by Laskov et al.
253 [21]. These results can be treated as “net specific oxygen release rate”. It should be
254 noted that the high result was obtained because the titanium (III) redox buffer was
255 adopted to determine the oxygen release. The titanium (III) redox buffer could
256 scavenge the oxygen as it was released, preventing it from being reabsorbed by the
257 respiring root tissue. When titanium citrate was used in a closed chamber, the rate of
258 root oxygen release was >80 times higher than in the nutrient solutions [22].

259 Root oxygen release rates obtained from two methods are depended largely on the
260 light intensity, which has exhibited a diurnal periodic variation. The daily changes of
261 oxygen release and light intensity followed unimodal distribution [16]. The diurnal
262 redox fluctuation as a result of photosynthetic activity of the plant may be a better
263 explanation for this phenomenon. Oxygen is produced through photosynthesis [23] and
264 is transferred from the leaves to the roots of plants via pressurized convective through
265 flow and molecular diffusion within the lacunal system of intercellular airspaces. The

266 pressure differential is generated by gradients in temperature and water vapour
267 pressure between the internal gas spaces and the surrounding atmosphere [24]. The
268 oxygen concentration is higher within the roots than the rhizosphere. Thus, oxygen is
269 released from roots, where reciprocating concentration gradients of oxygen. Presently,
270 there are several methods of the oxygen transfer measurement, such as microelectrode
271 measurement method [19, 20], oxygen-depleted solution and titanium (III) citrate
272 buffer [10, 22, 25] etc. However, the oxygen release from root systems in field could
273 be changed. Since the special rhizosphere environment in CW could affect the oxygen
274 release. Firstly, the moved wastewater in CW stimulated root oxygen release. The
275 oxygen released by root systems of plants could be easily transferred far away from
276 rhizosphere with increasing flow rate. Secondly, the root systems were growing in the
277 void of substrate. It affected the root morphology. Thirdly, the variations of dissolved
278 oxygen concentration in substrate water could affect the oxygen release. Therefore, the
279 mass balance method in this study could ideally describe the variations of oxygen
280 release in situ.

281 Another important fact for calculation the oxygen transfer into the rhizosphere is
282 the number of plants per unit of the surface and the mass of the roots. The number of
283 plants was reported about 3 plants/m² [26]. Dry weight of plant roots was averaged as
284 1.19g/plants. In this study, root oxygen release rate calculated by mass balance method
285 were slightly higher than the predicted values since the degradation processes of partial
286 organic and nitrogen compounds were mainly anaerobic (especially at night). However,
287 it was considered as aerobic process for the mass balance method.

288 **5. Conclusions**

289 Root oxygen release rate calculated by mass balance method is ranged from 20.3
290 to 58.3gO₂/m².d. The peak value of oxygen release rate (54.0-58.3gO₂/m².d) was
291 observed during the daytime at 15:00 hrs while the maximum light intensity was
292 observed at 13:00 hrs. The results showed that the rates of root oxygen release depend
293 largely on the light intensity, which was exhibited a diurnal periodic variation. The
294 daily changes of oxygen release and light intensity followed unimodal distribution.
295 Only minority (about 35.0% plus 8.7%) of the oxygen released by roots of wetland
296 plant were used in the degradation of organic matters and nitrogen-nitrification.
297 However, majority (55.8%) was used for roots respiration. Based on DO budget, root
298 oxygen release rate could be described by two fractions, including “net specific oxygen
299 release rate” and “dischargeable oxygen release rate”. This could help to explain the
300 debate about the quantity of oxygen released by root systems of wetland plant.

301

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306

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379 Fig. 1. Schematic description of the vertical subsurface flow CW system

380 Fig. 2. Diurnal fluctuation of oxygen consumption rate for organic matter degradation and

381 nitrification

382 Fig. 3. Diurnal fluctuation of DO concentration in substrate water, inflow and outflow

383 Fig. 4. Diurnal fluctuation of root respiration rate

384 Fig. 5. Diurnal fluctuation of root oxygen release rate and PAR

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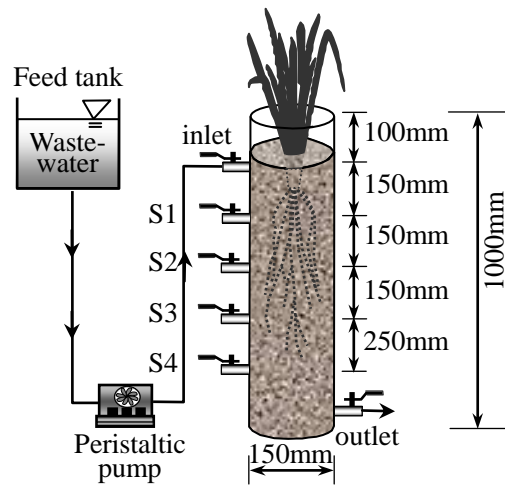


Fig. 1. Schematic description of the vertical subsurface flow CW system

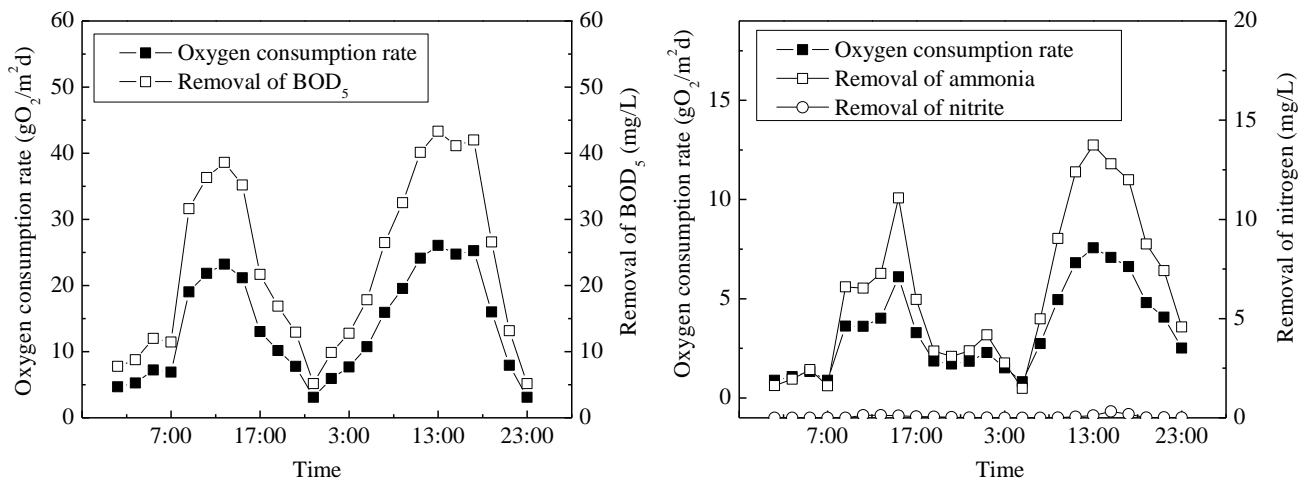


Fig. 2. Diurnal fluctuation of oxygen consumption rate for organic matter degradation and nitrification

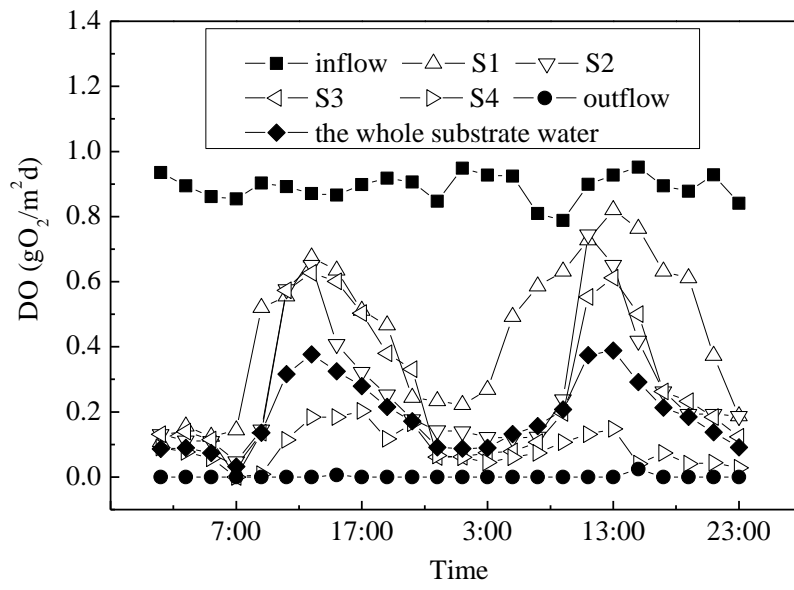


Fig. 3. Diurnal fluctuation of DO concentration in substrate water, inflow and outflow

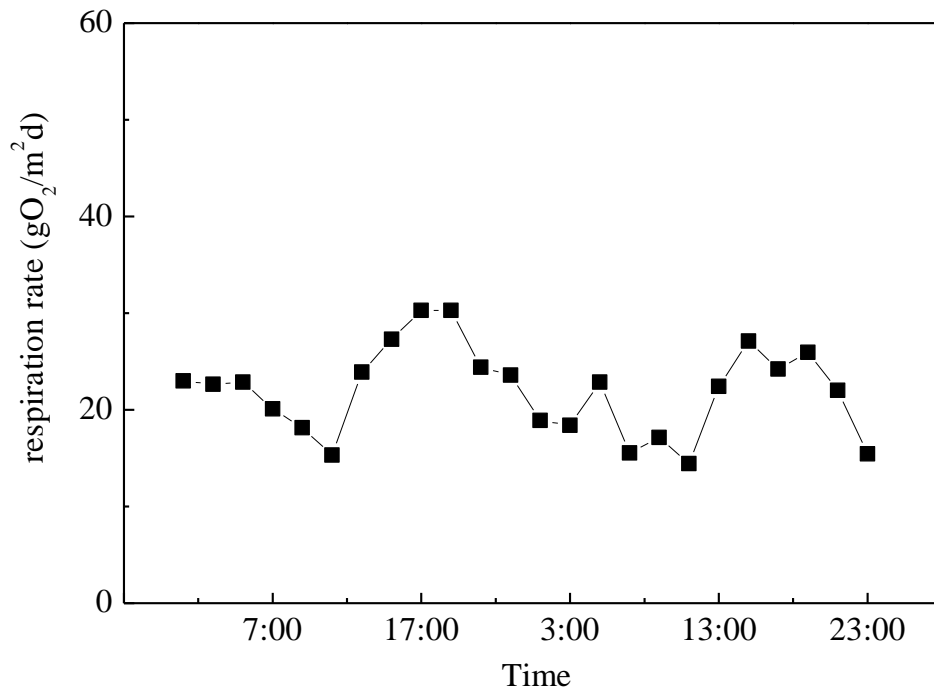


Fig. 4. Diurnal fluctuation of root respiration rate

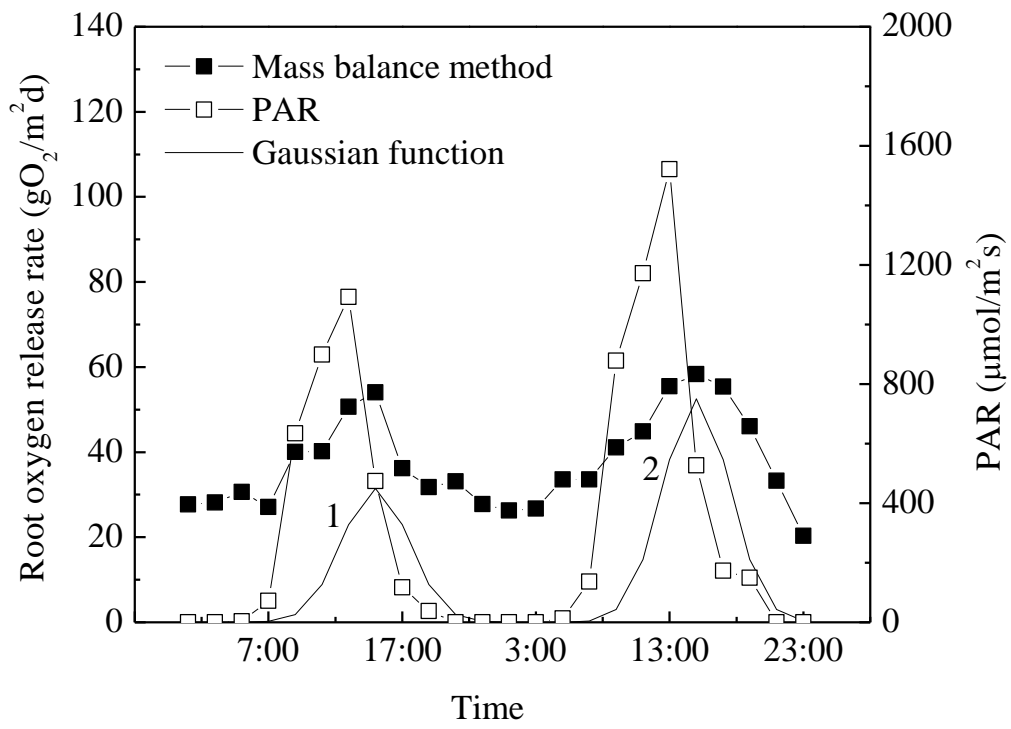


Fig. 5. Diurnal fluctuation of root oxygen release rate and PAR