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Bulk Metallic Glasses for Implantable Medical Devices and Surgical Tools

Philip Meagher, Eoin D. O’Cearbhaill, James H. Byrne, and David J. Browne*

With increasing knowledge of the materials science of bulk metallic glasses (BMGs) and improvements in their properties and processing, they have started to become candidate materials for biomedical devices. A dichotomy in the types of medical applications has also emerged, in which some families of BMGs are being developed for permanent devices whilst another family – of Mg-based alloys – is showing promise in bioabsorbable implants. The current status of these metallurgical and technological developments is summarized.

1. Introduction

Medical devices made from metallic alloys are in widespread use, and a process of continuous metallurgical improvement has led to enhanced biomechanical and biocompatible performance of these materials. The traditional alloy is a polycrystalline material, and a significant body of literature exists describing, for example, innovations in the formulation and processing of titanium- and cobalt-based alloys for orthopaedic applications.[3–7] Problems still being addressed include adverse corrosion, wear, and fatigue behavior, manufacturing difficulties, a lack of elasticity, and high modulus (which can cause stress shielding).

Bulk metallic glasses (BMGs) are a relatively new class of metallic materials developed over the past three decades. Whereas conventional alloys have a crystalline structure, BMGs exhibit no long-range atomic order, appearing instead as an atomically frozen liquid. In the last 20 years, many metallic alloys that can form glassy solids have been developed across a broad range of compositions, including Pd-, Pt-, Zr-, Fe-, Ti-, Mg-, Co-, and Au-based systems. The lack of atomic order yields elastic properties comparable to crystalline metals. [2] The glassy material is amorphous, which is made up of microstructures that extend up to micro-meter scales and, without grain boundaries or precipitates as oxidation sites, corrosion rates are significantly reduced compared to conventional metals.

BMGs soften above an alloy-specific glass-transition temperature \( T_g \) before eventual, time-dependent, crystallization at a higher, crystallization temperature \( T_x \). In this supercooled liquid region (SCLR) between \( T_g \) and \( T_x \), the viscous BMGs may be plastically shaped under low applied forces using polymer-processing techniques, such as hot embossing, extrusion, foaming, and injection and blow moulding, before again cooling to a solid metallic glass. With relatively low processing temperatures and no solidification shrinkage, parts can be formed to net-shape with excellent accuracy, and geometries, and surface patterns previously difficult in metals can be achieved with ease.[3–7]

Such remarkable properties and behavior have led to significant interest in BMGs as engineering materials over the past 20 years, but it is only recently that their potential for use as a biomaterial has been studied.[8]

To date, our understanding of material biocompatibility has evolved mainly through empirical testing, observing the interaction of materials with cells and host tissue in vitro and in vivo. Materials can induce host responses varying from local and systemic inflammation, hypersensitivitity, toxicity, and even tumorogenesis, meaning that thorough evidence of material safety is required before regulatory approval and clinical translation. We now largely recognise the material characteristics that generate both favorable and undesired host responses, as outlined by Williams,[9] offering the opportunity for informed material design, while being cognisant that biocompatibility is specific to the application and host environment.[10]

The original requirement of first generation “biocompatible” materials was bio-inertness.[11] This included a resistance to corrosion in the body, and here Ti- and Co-based alloys scored highly.[12] Less-inert metals, such as Mg, were therefore not deemed suitable, although the ions released on dissolution are generally not harmful to the human body. However, current design requirements for biomaterials, including most recently biometals, include eliciting an appropriate host response,[13] and this can include the need to biodegrade and resorb. There are potential advantages for BMGs that span applications of...
first-generation bioinert biomaterials through to third-genera-
tion materials that seek a controlled degradation profile and
interaction with the host.\textsuperscript{[14]} We therefore divide this article
on recent progress into applications of BMGs as: i) bioinert
devices, including permanent implantable devices and surgical
tools, and ii) biore(absorbable implants. It can be seen that the
alloy families are quite distinct in both cases.

2. Bioinert Systems

BMG alloys, which have potential for use as non-absorbable
medical devices, are usually based on Zr, Ti, or Fe. Toxic or
potentially harmful elements such as Be, Cu, and Al are gener-
ally to be avoided, if possible.

Certain BMG alloys have superior properties to current
biomaterials for applications as surgical tools or load-bearing
implants, such as high strength, low elastic modulus, and a
high elastic strain limit. The modulus provides a better match
to cortical bone, thus reducing stress shielding, which is useful
for hard-tissue prostheses.\textsuperscript{[12]} They also have very high hard-
ness and wear resistance. Toughness for some compositions
approaches values of crystalline metals, and is significantly
better than oxide glass/ceramics. Composites (crystals in an
amorphous matrix) rival crystalline counterparts for toughness
while retaining benefits of glassy structure, although few corro-
sion studies have been completed. Figure 1 shows the relative
strength and elastic stiffness of a range of BMGs from the liter-
ature,\textsuperscript{[12]} in comparison to Co–Cr, stainless steel, and Ti alloys.
BMGs are typically more corrosion resistant than the crystal-
line equivalents due to their chemical homogeneity and lack of
grain structure. The attributes of BMGs, which are particularly
suited to medical devices – both inert and resorbable – are sum-
marized in Table 1.

Specific BMG alloys have been under development/study for
medical applications. Due to the fact that nickel is toxic and a
possible carcinogen, the association of aluminium with Alzhei-
mer’s disease, and the biological toxicity of Be, Ni-free BMGs,
and compositions that are Al- and Be-free are desirable. However,
Be and Al both provide excellent glass-forming ability (GFA) and
were instrumental in development of BMGs, so not many early
BMG compositions exist without one of these elements.

2.1. Fe Systems

Wang et al. recently investigated the cor-
rosion resistance and biocompatibility of
three compositions of Fe-based BMGs,\textsuperscript{[16]} by
electrochemical measurements and indirect
contact cytotoxicity assays, respectively. In
comparison with 316 L SS biomedical stain-
less steel, Fe-based BMGs showed better
corrosion resistance in two simulated body
fluids (Hank’s solution and artificial saliva).
They found that a passive film on an Fe-based
BMG surface is quite stable, as with 316 L
SS; whereas Zr-based BMGs typically show
a weak repassivation ability in chloride envi-
ronments.\textsuperscript{[12]} The corrosion current densities
obtained from the anodic polarization curves
were lower for the BMGs than for the stain-
less steel. The analysis indicated that two of
the Fe-based BMGs have a larger polarization
resistance value than that of 316 L SS in arti-
ficial saliva. The pitting corrosion potentials
of Fe-based BMGs were much higher than

Table 1. Outlining the benefits, applications, and challenges associated
with the use of BMGs in medical-device applications.

<table>
<thead>
<tr>
<th>BMG attributes for medical devices</th>
<th>A. Bioinert devices</th>
<th>B. Biore(absorbable devices</th>
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<tr>
<td>High strength</td>
<td>Corrosion resistant</td>
<td>Optimal rate of corrosion and H\textsubscript{2} evolution</td>
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<tr>
<td>Formability in SCLR</td>
<td>Wear resistant</td>
<td>Strength decay inversely proportional to wound healing such as osteogenesis</td>
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<tr>
<td>Relatively low E</td>
<td>Controllable surface topography</td>
<td>Non-toxic ion release</td>
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<tr>
<td>High elastic strain limit</td>
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<td>Revision/removal surgery not needed</td>
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Potential Applications

- Orthopaedic prostheses
- Bone screw or surgical plate
- Surgical scalpels
- Intramedullary nails
- Flexible vascular stents
- Temporary vascular stents

Research Challenges

- Unknown fatigue, stress corrosion, and wear debris behaviour
- Matching GFA with clinical needs and biocompatibility

Figure 1. Range of mechanical properties of metallic biomaterials. Data for the information in
the figure are taken from refs. [12], [13], and [15].
that of the 316 L SS, resulting in very few ions being released into the electrolytes, while a significant level of Ni- and Fe-ion release was found for 316 L SS under the same conditions. The indirect cytotoxicity results suggest that all three Fe-based BMG extracts have no cytotoxicity to L929 and NIH3T3 cells. Fe-based BMGs therefore show promise for biomedical applications, especially in dental implantology, due to high toughness and wear resistance, and where their limited GFA and higher elastic moduli would pose less of a problem.

2.2. Ti Systems

Recent work has focussed on development of BMG alloys with suitable properties, whilst avoiding toxic components. Attempts by Zhu et al. showed good GFA with $d_{\text{crit}}$ up to 7 mm, and no Ni, Be, or Al, but did include significant amounts of Pd.\cite{17} The corrosion behavior of the alloys in both lactic acid and PBS was excellent, offering significant resistance to pitting over both pure Ti and crystalline Ti-6Al-4V. The strength was reported to be 1970 MPa, with significant ductility in ribbon samples, although bulk ductility is not mentioned.\cite{18} Cu-free alloys were produced by Oak and Inoue in the Ti–Pd–Zr–Si system – with good corrosion resistance, high hardness and strength, and good ductility, although the samples were in ribbon form only due to poor GFA.\cite{19} More recently, alloys designed based on individual elemental toxicity, were produced in fully glassy form by Calin et al.\cite{20} in the TiNbZrSi system, although again only in ribbon form – bulk samples were not possible due to the low GFA – but the corrosion behavior was again superior to Ti-6Al-4V. Xie et al. reported that spark sintering of gas-atomized Ti-MG powders can produce high-density, large-volume parts\cite{21} with good corrosion resistance (although no comparisons were made to as-cast BMGs or commercial alloys). Lin et al. very recently reported interesting results for a TiZrTaSi alloy, which contains no toxic elements, showing improved mechanical properties over the previously mentioned alloys, corrosion resistance much higher than that of Ti-6Al-4V and 316L SS, some ductility, and a very wide supercooled-liquid region, which suggests that the alloy could be formed in bulk.\cite{22} Pang et al. continued in this vein to develop a TiCuZrFe5SnSiAg system with a critical casting diameter up to 7 mm. The material exhibited strength of 2 GPa, significant compressive bulk plasticity, similar corrosion resistance to that of Ti-6Al-4V in PBS, and similar cell viabilities with MC3T3-E1 pre-osteoblast cells.\cite{23}

2.3. Zr Systems

Liu et al.\cite{24} and Sun et al.\cite{25} performed in vitro corrosion and cytotoxicity testing on ZrCuAlAg alloy systems, all of which exhibit extremely good GFA (up to 20 mm $d_{\text{crit}}$). The Ag-bearing BMGs performed significantly better than Ti6Al4V in Hanks solution\cite{26} and phosphate-buffered saline (PBS)\cite{27} due to AlO$_3$ passivation (which appears to be increased by Ag content), and showed marginally improved cell viabilities with L929\cite{28} cells, and marginally worse with MG63\cite{29} cells over Ti6V4Al, falling into the same cytotoxicity Grade 0–1 (along with all current biomedical alloys). Ag-bearing BMGs also show antibiotic effects,\cite{30} and could lead to a reduction in surgical revisions due to infection. In vivo implanted BMGs showed excellent host compatibility, with no gas formation nor immune response, and show similar osseointegration as Ti6Al4V.\cite{31}

The Ag-bearing Zr alloys exhibited\cite{24} fatigue strength in excess of 400 MPa (10$^7$ cycles) and high fracture toughness, as well as some compressive plasticity up to 0.2%. With a strength of ca. 1800 MPa and hardness around 430 GPa, the BMGs compare favorably with current biomedical alloys.\cite{25} In the same research group, Wang et al. carried out fatigue testing on a similar alloy composition in SBF, whereby the fatigue limit decreased only by about 10%, to 366 MPa.\cite{27}

Lin et al.\cite{28} tested 7 Zr- & Ti- based metallic glasses in Hanks solution and human serum. Excessive copper, above 17.5 at%, rapidly increased the electrochemical activity and sample corrosion. While bulk samples showed no cytotoxicity, the ionic corrosion products of the high-Cu alloys showed acute toxicity to D1 bone-marrow stem cells.\cite{28}

Hua et al. compared a low-Cu Zr$_{65}$Ti$_{12.5}$Fe$_{7.5}$Cu$_{10}$Ag$_5$ MG with Ti-6Al-4V biomedical alloy for biocompatibility using MC3T3-E1 mouse pre-osteoblast cells, with excellent cell attachment,\cite{32} and improved corrosion performance in PBS over the alloys reported in ref.\cite{24} They further optimized the BMG composition to increase the toughness and bulk compressive plasticity up to 12%, while maintaining impressive strength. The paper does not mention $d_{\text{crit}}$, but a high value is expected due to the large SCLR of 62 K.\cite{29} When immersed in an osteogenic medium, bone matrix deposition by attached osteoblasts occurred at a significantly higher rate on an Al-bearing Zr BMG over Ti6Al4V.\cite{33} Other researchers achieved a strongly bonded, bioactive calcium titanate coating on the surface of a Zr BMG by laser cladding, suggesting this would lead to greatly enhance tissue adhesion and growth on the material.\cite{30}

3. Specific Bioinert Device Applications

3.1. Surgical Tools (Scalpels)

Due to the lack of any grain-size limitations on feature size, very sharp edges are possible on BMGs. Sharpness improvement on surgical blades by using BMG and metallic glass coatings has been reported,\cite{34} low values of blade sharpness index (b.s.i.) indicate sharper edges. The b.s.i. on a fabricated ZrCuAlAgSi BMG blade was 0.24, a value of 0.23 was achieved for a martensitic steel blade coated with a thin film of the same alloy, whereas the uncoated steel blade value was 0.34. A surgical blade coated with an Fe-based metallic glass thin film, demonstrated a b.s.i. value of 0.28, while also displaying a significant increase (65%) in blade durability, due to its extreme hardness (1200 Hv).\cite{35} The local roughness of all coated blades was extremely low (<5 nm).

Chu et al. sputter-coated a single-crystal Si wafer blade with Zr- and Cu-based thin-film metallic glasses,\cite{36} thereby achieving a 27% increase in blade sharpness. They also reported significantly increased durability of blades throughout testing with no blade edge defects developing, in comparison with an uncoated blade, and significantly improved surface roughness.
on coated blades. Good antimicrobial activity against *Escherichia coli* and *Staphylococcus aureus* was achieved, as shown in Figure 2. Improved antimicrobial activity, against these and other common pathogens, on stainless steel sputter-coated with a thin film of Zr-based metallic glass, when compared with uncoated stainless steel, has also been reported.[34]

### 3.2. Stents

BMGs are also being evaluated as potential cardiovascular stent materials, and finite element modelling has been used to predict the relevant mechanical behavior of a ZrAlFeCu BMG in vivo.[35] The alloy, developed to avoid Ni and Be, has shown improved hardness and strength[35] over 316L SS (roughly twice, in both cases), which allows for stent struts to be markedly thinner – shown to significantly reduce restenosis[36] and improve the deliverability of the device. The FE results (see Figure 3) showed that the high elastic limit and low modulus allow a stent to flex more (facilitated by the thinner members) and deform during heart beats. Comparisons of the mechanical properties between stented and unstented arteries showed a better match over a 316L stent – compliance mismatch has been shown to be correlated with intimal hyperplasia, which also leads to restenosis.[37] Corrosion resistance in PBS was significantly improved over 316L SS in vitro, due to the formation of a ZrO$_2$-rich oxide film,[35] thus overcoming the relatively poor results that were summarized in an earlier report[12] for different compositions. Improved cell adhesion over that on 316L SS was observed, together with slower growth of smooth muscle cells, again reducing the potential incidence of restenosis. However, a corrosion fatigue study was not carried out.

### 3.3. Other Applications

Researchers showed[38] that ZrAlCuNi BMG offers similar performance to Ti6Al4V, when used as stabilizing intramedullary nails in bone fixation in vivo in rat femurs, although with reduced bone-nail bonding, leading to easier removal when healing is completed.

The Ni-free Zr BMG used by Liu et al.[24] exhibited a 20-fold decrease in wear rate of ultrahigh-molecular-weight polyethylene (UHMWPE) acetabular cup material when compared against the performance of a CoCrMo biomedical alloy in both Hank’s solution and calf bovine serum, due to its highly hydrophilic surfaces increasing fluid lubrication. The wear results are reportedly superior to any current conventional metal-on-polymer joints.[19] Wear on the alloy itself shows significant improvement over Ti6Al4V in dry conditions, and in both PBS[40] and foetal bovine serum.[41]

Foam of BMGs match the moduli and yield strength/elasticity of human bones very well. These BMG foams (BMGFs) are promising for bioimplant application without a significant stress-shielding effect, and the porosity allows for bone ingrowth and vascularization.

Schroers et al.[42] first reported BMG foams in a Pd system, created via high-pressure alloying with B$_2$O$_3$ and then heating the BMG into the SCLR in a low-pressure environment. The created foam showed 88% porosity, and exhibited excellent matching with the properties of cancellous bone, in terms of stiffness and compressive strength. Due to the size-dependent nature of the mechanical properties in BMGs,[43] the foam's thin struts demonstrate excellent plasticity.[42] Li et al. also demonstrated BMG foams fabricated via hot-pressing of BMG powders with NaCl, and subsequent dissolution of the NaCl to reveal an open porous foam. A Young’s modulus as low...
as 10 GPa was reported for a non-biomedical Zr-based BMG system (ZrCuNiAl).[44]

4. Bioresorbable Systems

4.1. Magnesium as a Biomaterial

There are clinical advantages for certain metallic medical implants to dissolve in a controlled fashion in vivo following surgery, provided the corrosion products are not harmful, thus obviating the need for further surgery to extract the implant after it has served its function, reducing both the burden on the health system, and the associated risks toward the patient.[15]

Ever since the first reported use as wire ligature in 1878,[45] magnesium has been suggested for biomedical applications. Found in natural abundance in vivo, it plays an important biological role in many metabolic functions, and, as a result, is extremely biocompatible, with the average 70 kg adult carrying approximately 25 g, primarily in bone tissue.[46]

One of its conventional engineering weaknesses, namely its poor corrosion resistance in electrolytic environments, suddenly becomes a potentially attractive strength – as it degrades in vivo to a harmless oxide that is efficiently excreted by the body.[46] Interestingly, the corrosion products of magnesium metal exhibit a remarkable antibiotic effect against a number of common aerobic bacterial organisms. Measured in vitro, Mg²⁺ ions offer an equivalent antibacterial effect to fluoroquinolone by increasing local pH values, and, if effective in vivo, could greatly reduce the need for revision surgeries due to infection.[47]

As an absorbable stent, magnesium would greatly reduce the risk of restenosis, chronic vessel inflammation, and patient injuries associated with permanent stents, with improved mechanical behavior and cost over current absorbable polymer materials.[48,49] As a potential bone-fixturing material,
magnesium alloys exhibit exceptionally low densities of 1.7–2 g cm\(^{-3}\); significantly lower elastic moduli than current titanium and cobalt–chrome biomedical metals, and strength and toughness exceeding polymer and ceramic biomaterials respectively, as outlined in Table 2. Furthermore, implanted magnesium has been shown to provide a stimulatory effect on the rate of bone regrowth in the surrounding tissue.\[50,51\] Indeed, a drug-eluting, absorbable magnesium stent, and absorbable bone screws are now commercially available, and have both been the recent subjects of positive clinical human trials.\[52,53\]

### 4.2. Glassy Magnesium Alloys

Despite the recent resurgence in interest and limited commercial success, conventional crystalline magnesium alloys still exhibit significant issues. Strength and fatigue values have been reported to decrease by an order of magnitude under an applied strain in vitro.\[54\] Despite continuing research, too-rapid corrosion can lead to implant failure before healing can complete, and the same corrosion often leads to catastrophic H\(_2\) evolution in patient areas with poor transport mechanisms.\[59\] Magnesium-based BMGs exhibit remarkably improved mechanical properties when compared with crystalline equivalents, as per Table 2. Indeed, a number of Mg-based BMG families exhibit high strengths, often exceeding 1000 MPa, greater than stainless-steel biomedical alloys,\[57\] and, in some cases, even titanium and cobalt–chrome alloys.\[58\]

Without initiation sites for intergranular attack, glassy Mg alloys have significantly enhanced corrosion resistance over their crystalline forms.\[59\] Corrosion resistance of magnesium is also heavily influenced by alloying elements in solid solutions,\[60\] but the solubility of most alloying elements is relatively low. With no solid solubility limits, metallic glasses offer the potential to greatly improve corrosion rates.

While studies into magnesium BMGs for engineering applications have gathered momentum over the past 30 years due to their light weight and excellent mechanical properties, the majority of Mg-based compositions studied contain significant copper, nickel, and/or gadolinium, and hence are unsuitable for use in absorbable applications. It is only since 2008 that research has focused specifically on the Mg–Zn–Ca family, as a biologically compatible system with potential for medical devices.\[61\]

Glassy, Mg-rich alloys with a Zn content above 29 at%, show a marked increase in corrosion resistance over any current crystalline Mg alloys.\[62\] The composition develops a dense, amorphous ZnO/ZnCO\(_3\) layer on the surface, preventing the rapid dissolution of the bulk magnesium. H\(_2\) gas evolution is effectively suppressed in vivo, without any effect on the bio-compatibility, and the ionic degradation products remain well within the recommended daily allowances for humans.\[63\] Such behavior in not possible with crystalline Mg, as the Zn content would form secondary intermetallic phases rather than a homogeneous structure, and the internal galvanic couple would accelerate corrosion.\[64\]

MG\(_{66}\)Zn\(_{30}\)Ca\(_{4}\), for example, corrodes in vivo at less than half the rate of WE43, a particularly corrosion resistant crystalline alloy, and at less than 20% the rate of other common magnesium alloys, outlined in Figure 4. Exhibiting a strength of roughly 800 MPa in both tension and compression, a high elastic strain up to 2%, and high fracture toughness, the alloy shows significantly improved mechanical properties over conventional Mg alloys. An elastic modulus of 45 GPa also reveals a stiffness close to that of cortical bone.\[55, 56, 64–68\] Microalloying with 2 at% silver or strontium in place of calcium further enhances the corrosion properties and strength of the alloy, as well as improving the antibacterial effect (Ag) and osteogenesis (Sr).\[69, 70\] Cytotoxicity testing revealed that alloy samples were able to support significant cellular activity to a high degree against a control, with increased cell viabilities up to 90%, (compared to 60% for crystalline Mg alloys), while a stimulatory effect on the rate of bone growth is also reported.\[66, 67, 71, 72\]

Despite such promising characteristics, the alloys must overcome some critical stumbling blocks to be seriously considered for clinical purposes. Critical casting diameters of the initial MgZnCa alloys were capped at 3 mm, potentially restricting use to small pins and screws. As-cast ingots of higher diameters are only partially amorphous.\[74\] More recently, small alloying additions of Sr have been shown to enable an amorphous structure up to 6 mm.\[69\] Further exploration of the composition space revealed Ca-rich alloys with significantly increased casting diameters, and improved thermoplastic processing properties, excellent cell proliferation, and increased bone hyperplasia.\[72, 75, 76\] While the Young’s modulus of the ternary systems is almost identical to that of dense cortical bone (20–35 GPa), the strengths are not as impressive as the Mg-rich systems, exhibiting failure in compression at around 400 MPa. Despite very limited H\(_2\) evolution, the corrosion of the ternary alloys is very rapid, with some of the Ca-rich alloy pins dissolving completely within just 3 h.\[75\] Recently however, it has been demonstrated that additional alloying elements such as Yb and Sr can significantly reduce the corrosion rate of the Ca-rich systems (Figure 4), bringing it significantly below crystalline Mg-alloys, and suggesting a viable alternative to the smaller-diameter Mg-rich systems.\[73, 77\]

Moreover, while ductility and reasonable flaw tolerance are reported in sub-millimeter Mg-BMG samples,\[86\] bulk samples...
of the ternary alloy still exhibit the lack of plasticity characteristic of many metallic glasses. Initial surface corrosion offers plentiful crack initiation sites, which can propagate rapidly through the bulk of the material, causing brittle failure.[78] Interestingly, due to rapid bone ingrowth into the samples, the cracking does not appear to cause catastrophic failure of the implants in vivo, and a recent comparative study reports that despite the failures, the implanted glassy alloys performed better than both PHB polymer and crystalline MGWZ21 implants.[79]

In an attempt to address the typically brittle nature of Mg BMGs, Yb addition at 2 at% was found to introduce significant bending ductility, while excellent corrosion and biocompatibility properties were retained.[68] In a different approach, alloying with Y at 10 at% is shown to produce a composite material, with crystallites of roughly 10 μm embedded in a glassy matrix. The material exhibits 19% plasticity and ductile failure, while retaining the strength and elasticity of a BMG, and corrosion rates only marginally higher than a fully glassy component.[80] However, there remains a dearth of research on the mechanical behavior of all the Mg BMG systems in a corrosive biological environment, and further work to understand their performance in both in vitro and in vivo is required before serious clinical consideration will take place.

5. Conclusions

Progress in development of medical devices is often materials-limited, and new bulk metallic glasses are enabling new types of implant and surgical tools. Bioinert alloys, based on Ti, Zr, or Fe, are being developed for potential use as cardiovascular stents, permanent orthopaedic prostheses, and surgical devices such as scalpels. Research on bioabsorbable BMGs based on Mg is also yielding promising results, with the ultimate goal being implants that do not need a follow-up surgical intervention for their removal. In all cases the applications will rely on the unique properties of amorphous metals, but significant scientific and engineering challenges need to be overcome before the successful clinical applications of most of the relevant medical devices.